YESAYAN, G.T.; GALOYAN, G.A.; BABAYAN, A.A.; POSTOYAN, N.R.

Interaction of sulfochlrides with dimedon. Dokl. AN Arm. SSR 38 no.5:301-304 '64. (MIRA 17:6)

1. Institut organicheskoy khimii AN Armyanskoy SSR. Predstavlenc akademikom AN Armyanskoy SSR V.I.Isagul yantsem.

YESAYAN, G.T.; OGANESYAN, E.Ye.; ASOYAN, E.L.

Transformations of disulfonyl chlorides. Part 1: Interaction of alkanedisulfonyl chlorides with phenols and aromatic amines containing a halogen and a nitro group. Izv.AN Arm.SSR. Khim. nauki 17 no. 3:339-344 '64. (MIRA 17:7)

1. Institut organicheskoy khimii AN Armyanskoy SSR.

YESAYAN, G.T.; OGANESYAN, F.Ye.; ASOYAN, E.L.

Transformations of disulfuryl chlorides. Part 2: Synthesis of 4-methyl-7-coumaryl and 8-quinolyl esters of some disulfo soids. Izv. AN Arm. 99R. Khim. nauki 18 no.3:309-312 65.

(MIRA 18:11)

1. Institut organicheskoy khimii AN ArmSSR. Submitted May 15, 1964.

USSR/Cultivated Plants. Fruits. Berries.

11

Abs Jour : Ref Zhur-Biol., No 15, 1953, 68324

Author Inst Yesayan, G. Ye.

Armenian Scientific Research Institute of Viniculture, Wine Production, and Fructicul-

ture.

Title

: Research Results of Agricultural Engineering Techniques Applied for Fruits.

Orig Pub: Byul. nauchno-tekhn. inform. Arm. n.-i. in-tavinogradarstva, vinodeliya i plodovolstva,

1957, No 1, 41-43

Abstract: Detween 1953 and 1955, the Armenian Scientific Research Institute of Vinboulture, Wine Production, and Fructiculture conducted some research

which was aimed at finding a system for main-

Card : 1/2

154

### YESAYAN, M.A.

State of the coronary circulation in resumatic fever. Trudy Inst. klin. i eksper. kard. AN Gruz. SSR 8:481-486 [63. (MIRA 17:7)

1. Institut kardiologii AN Armyanskoy SSR, Yerevan.

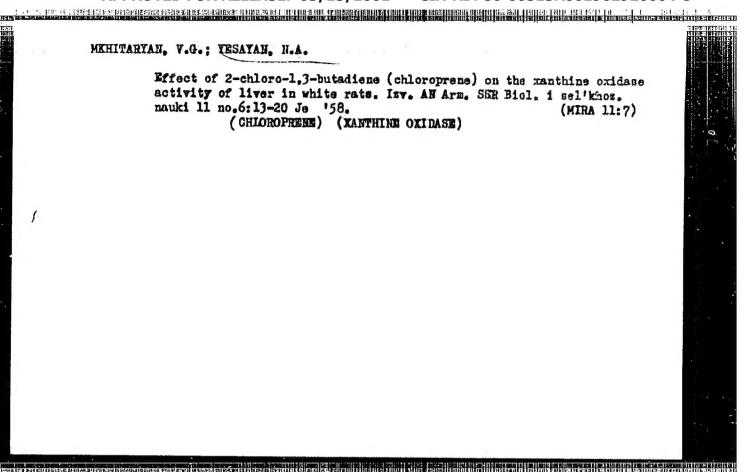
YESAYAN, N.A., nl.nauchn. sotrudnik

Hemodynamic changes in neurogenic cardiopathies. Vop.kardiol.

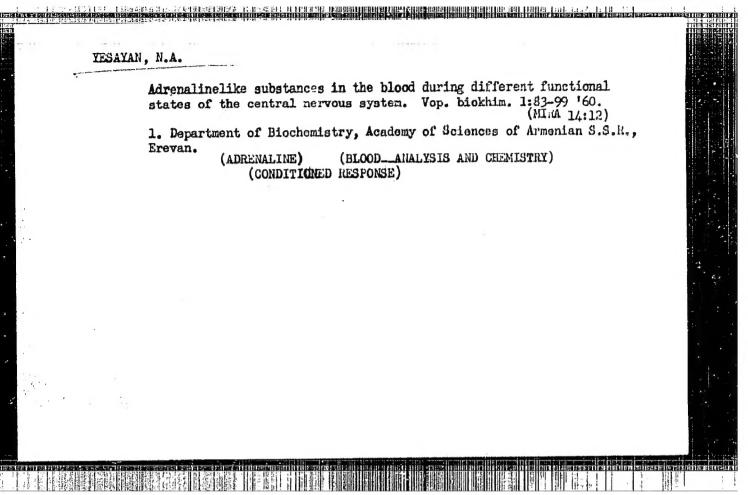
no.1:156-161 '56. (MIRA 12:9)

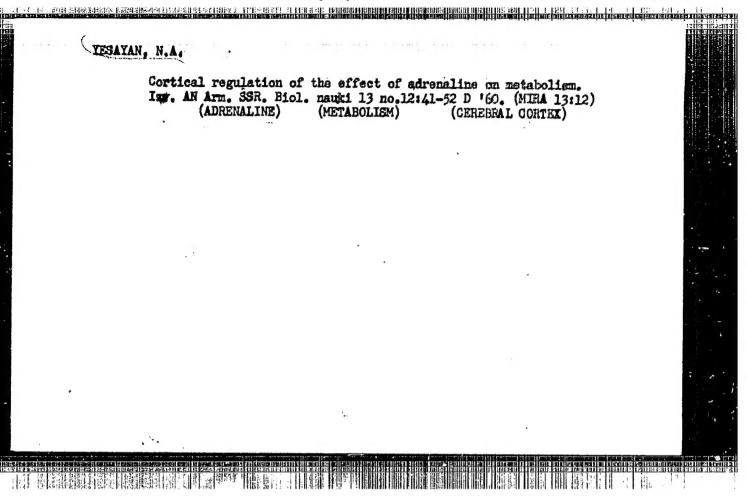
1. I2 Sektora meditsiny AN Armyanskoy SSR.

(CARDIOVASCULAR SYSTEM--DISMASES) (BLOOD)



# TESAYAN, N.A. Changes in the amount of adrenalinlike substances, histamine and glunose in blood during conditioned adremalin reflexes and internal inhibition (in Armenian with summary in Runslan). INV.AN Arm. SSR.Biol. i sel'khos.nauki 11 no.11:55-66 M '58. (MIRA 11:12) (ADREMALIN) (BLOOD—ANALYSIS AND CHEMISTRY) (CONDITIONED RESPONSE)

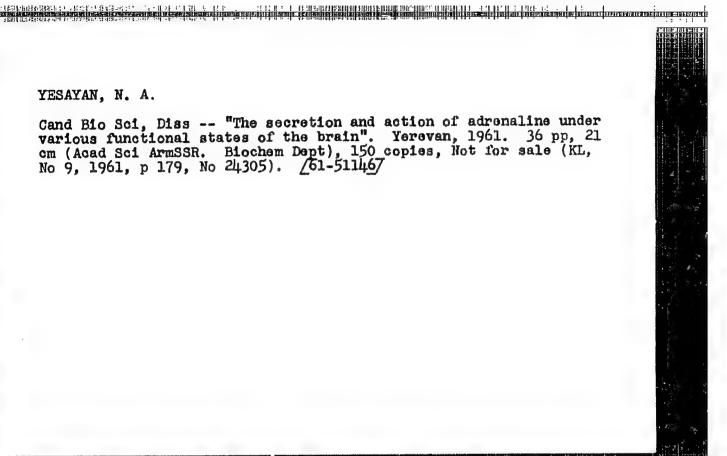


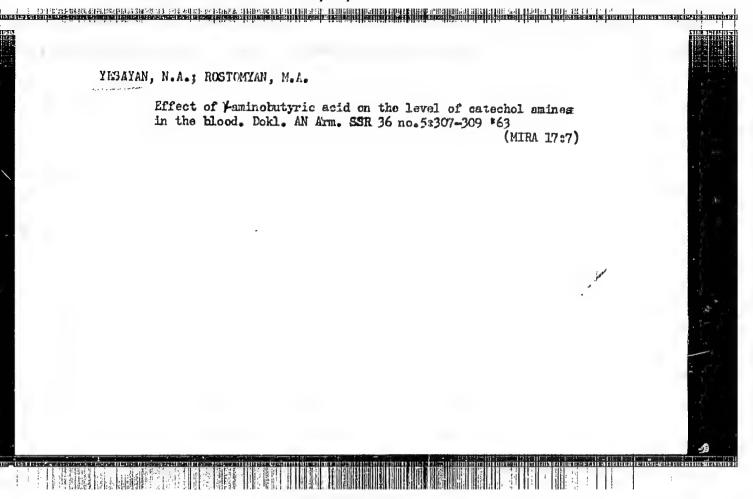


YESAYAN, N.A. (USSR)

"Cortical Regulations of the Secretion and Function of Adrenaline."

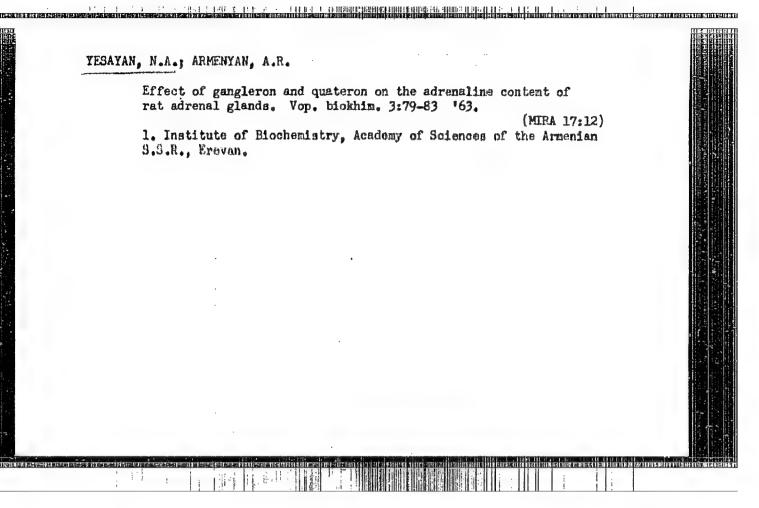
Report presented at the 5th Int'l. Biochemistry Congress, Woscow, 10-16 Aug 1961.

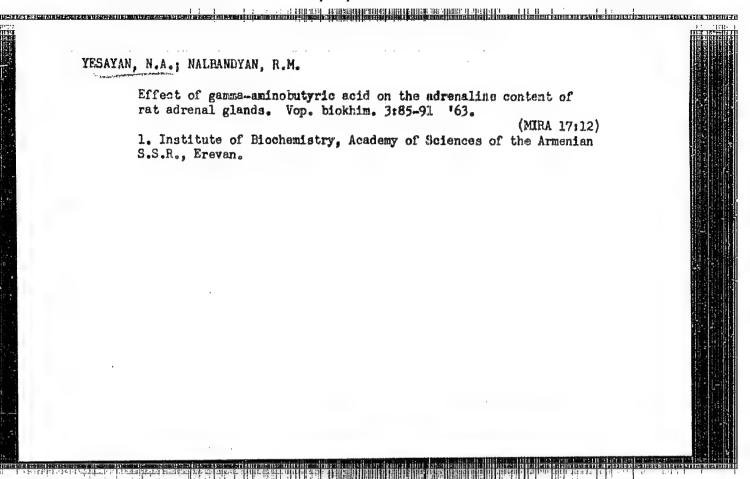




YESAYAN, N.A.; ROSTOFYAN, M.A.

Adrenalinelike substances in the blood during a conditioned pain reflex and internal inhibition, Izv. AN Arm. SSR. Biol. nauki 16 no.3:35-44 Pr 163. (MRA 17:10)





YESAYAN, N.A.; KAZAROVA, Ye.K.

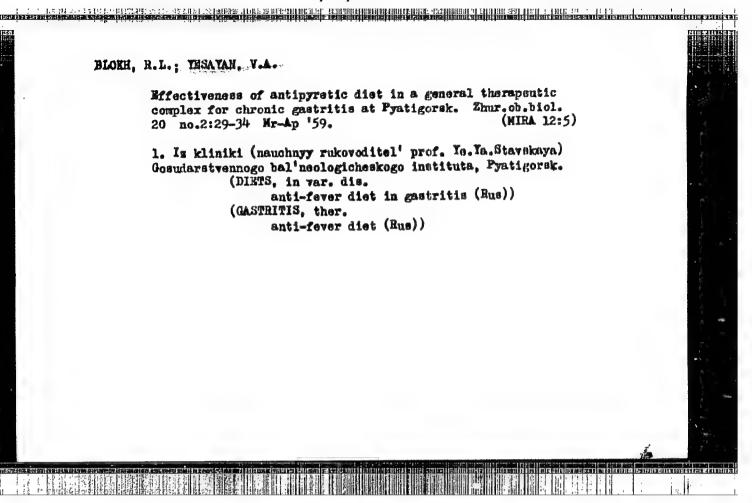
Effect of gangleron on the garma aminobutyric acid content in the brain. Yop. mickhim. moz. 1:67-72 '64. (MERA 18:9)

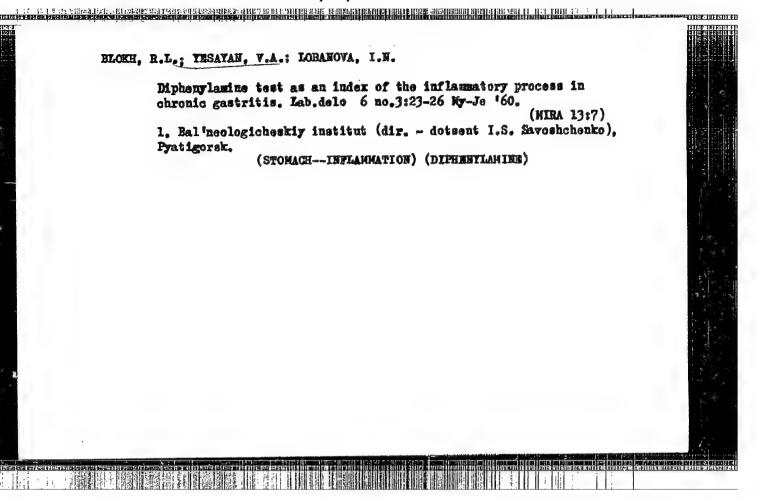
1. Institut biokhimii AN Arrasa, Yerevan.

YESAYAN, N.A.; ARMINYAN, A.R.

Effect of dopamins on the absorption of glucose by ret brain sections. Vop. biokhim. moz. 1;123-130 \*64. (MIRA 18;9)

1. Institut biokhimid AN ArmiSR.





VOROB'YEV, P.I.; YESAYAN, Ye.R.; RYAHOV, Ye.I.

IAkov Alekseevich Vlasov; October 22, 1900 - November 5, 1963.
Pochvovedenie no.5:119 My '64.

(MIRA 17:9)

### YESAYULENKO, P. I.

"The Effect of the Sowing Period on the Growth, Development, and Yield of Table Root Plants in Alma-Ata Suburban Areas." Cand Agr Sci, Kazakh Agricultural Inst, Min Higher Education, Alma-Ata, 1955 (KL, No 9, Feb 55)

SO: Sum. No 631, 26 Aug 55-Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (14)

YESBERG, N.A.; SHATALOV, N.N., nachal'nik; RPSHTEYN, G.Ya., professor, starshiy khirurg.

Tissue therapy in certain diseases. Vest.khir. 73 no.4:55-56 JI-Ag \*53.
(MRA 6:8)

1. Leningradskiy gorodskoy gospital' dlya lecheniya invalidov Otechestvennoy voyny.

(Tissue extracts)

YESDANYAH, B.A.; MANVELYAN, K.R.; KUMKUMADZHYAN, V.A.

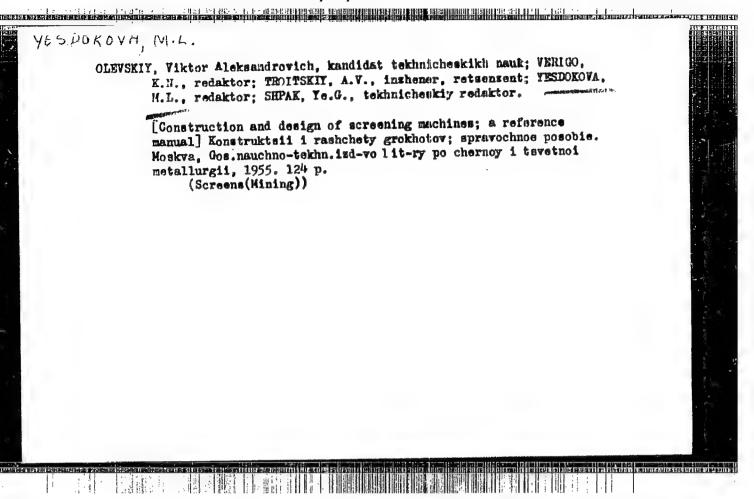
Morphological and histochemical data on Ehrlich's cascingua following its treatment with some preparations. Izv. AN Arm. SSR. Biol. nauki 18 no.5:44-51 My '65. (MIRA 18:7)

1. Institut rentgenologii i onkologii AMNI SSSR.

DOLININ, G.A.; STEPANYAN, A.N., veter. vrach.; YESHCHENKO, N.A.; OREKHOVSKIY, V.K.; LYSENKO, I.F., veter. vrach (Tiraspol' Moldavskoy SSR); SAPAYKIN, I.M., prof.: POGULYAY, V.D., veter. vrach (Romanovskiy rayon, Altayskogo kraya); BOGDANOVSKIY, A.V.; SAVUSHKINA, Ye.T., kand. veter. nauk

Prophylaxis and treatment of dyspepsia in calves. Veterinariia 41 no.1:72-75 Ja 164. (MIRA 17:3)

1. Glavnyy veterinarnyy vrach sela Uren', Gor'kovskoy oblasti (for Dolinin). 2. Ivanovskaya mezhrayonnaya veterinarnaya laboratoriya Khersonskoy oblasti (for Stepanyan). 3. Starshiy veterinarnyy vrach sovkhoza "Kamenskiy" Moldavskoy SSR (for Saraykin). 4. Moldavskiy sel'skokhozyaystvennyy institut (for Saraykin). 5. Glavnyy veterinarnyy vrach sovkhoza "Berestovoy", Donetskoy oblasti (for Bogdanovskiy).



15(2) SCV/72-59-4-6/21 AUTHORS: Krasnovskiy, O. V., Bil'tyukova, E. P., Yecechko, A. M. TITLE: Complexometric Determination of Calcium Oxide and Magnesium Oxide in Vertically Drawn Glass (Kompleksometricheskoye opredeleniye okisi kal'tsiya i okisi magniya v stekle vertikal'nogo vytyagivaniya) PERIODICAL: Steklo i keramika, 1959, Hr 4, pp 22 - 25 (USDR) ABSTRACT: The complexometric method of titration makes it wossible to simplify considerably the accelerated glass analytis (sec paper by O. V. Krasnovskiy, Ref 1). The determinations of calcium and magnesium may be carried out according to two methods as may be seen from the papers by K. B. Yatsimirskiy on the one hand and by T. B. Styunkel! and Ye. M. Yakimets on the other (Refs 2 and 3). In order to check the complexometric methods of determination comparative investigations with synthetic solutions were carried out, the salt content of which corresponded to those of the solutions in the glass analysis. In table 1 the experimental results are given. Later, the same experiments were carried out with Card 1/2 samples of industrial sheet glass which was drawn vertically.

THE OFFICE AND ADDRESS OF THE STATE OF THE S

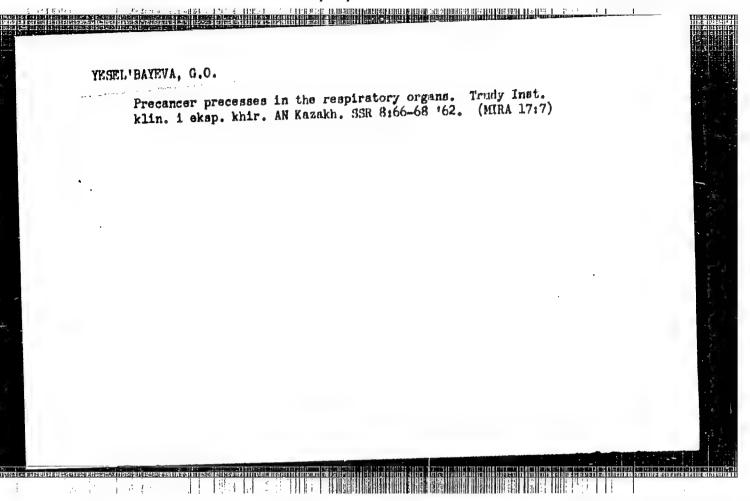
Complexometric Determination of Calcium Oxide and SOV/72-59-4-6/21 Magnesium Oxide in Vertically Drawn Glass

The results of these analyses are listed in table 2. In conclusion, 2 formulae are mentioned according to which the per cent content of CaO and MgO in glass may be computed. The necessary specific reagents and their preparation are shown in the "Instructions for the Determination of Water Hardness by Means of the Complexometric Method" (MKhP, SSSR, 1957, pp 1-5). There are 2 tables and 3 Soviet references.

ASSOCIATION:

Institut stekla (Glass Institute)

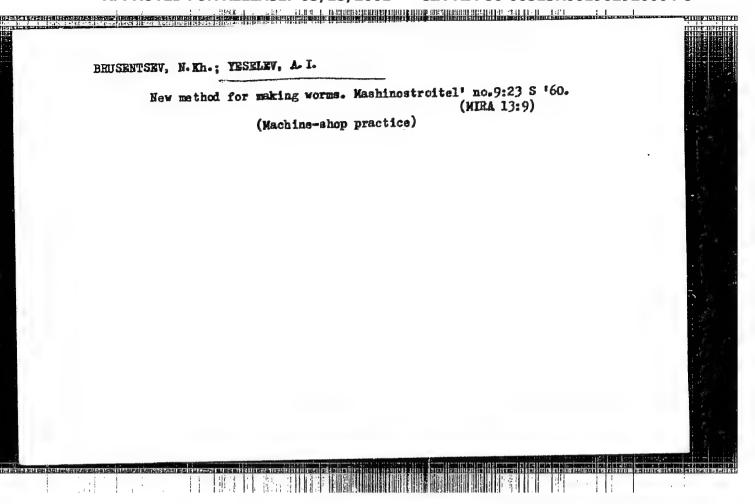
Card 2/2

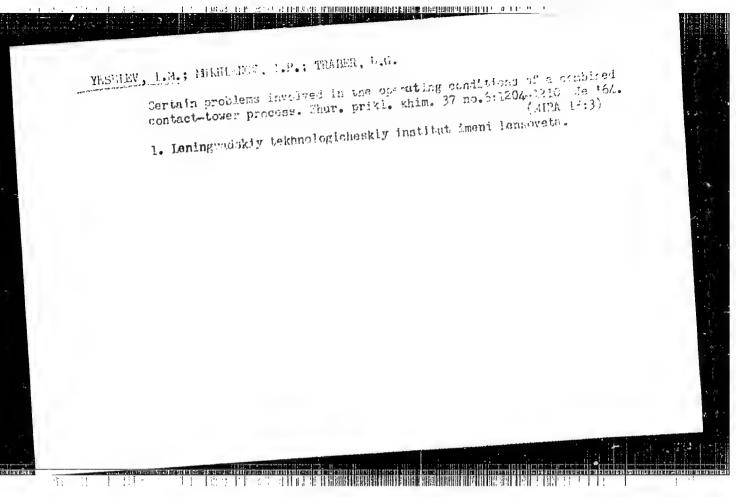


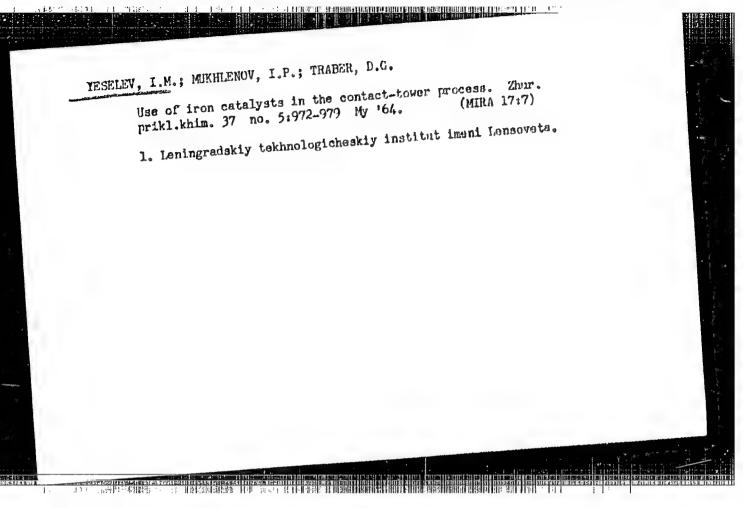
YESEL'BAYEVA, R.

An Outstanding athematican of St. PETERSBURG p. 80

TRANSACTIONS OF THE 2ND REPUBLICAN CONFERENCE ON MATHEMATIKES AND MECHANICS (TRUDY VIORGY RESPUBLIKANSMOY KONFERENTSII PO MATHMATIKE & MEKHANIKE), 18% pages, published by the Publishing House of the AS KAZAKH SSR, ALMA-AMA, COCE, 1962







YESELEV, I.M.; MUKHLENOV, I.P.; TRABER, D.G.

Use of an iron catalyst in the contact-tower process. Zhur. prikl. khim. 37 no. 4:722-727 Ap '64. (MIRA 17:5)

1. Leningradskiy tekhnologicheskiy institut imeni Lensoveta.

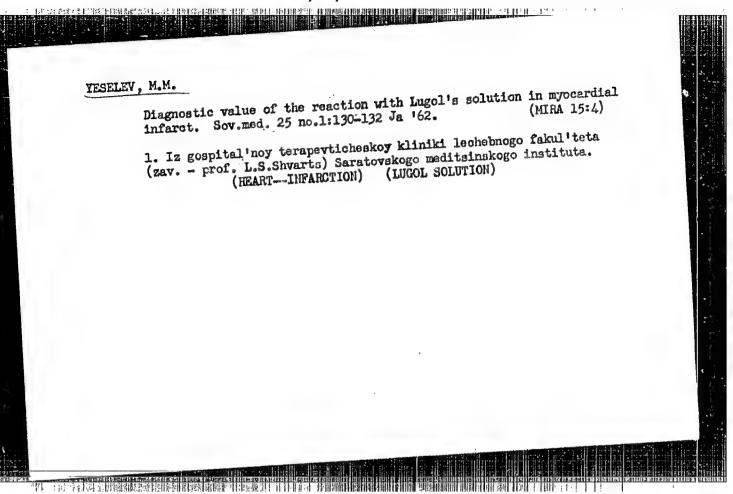
LARINA, V.S.; YESELEV, M.M.

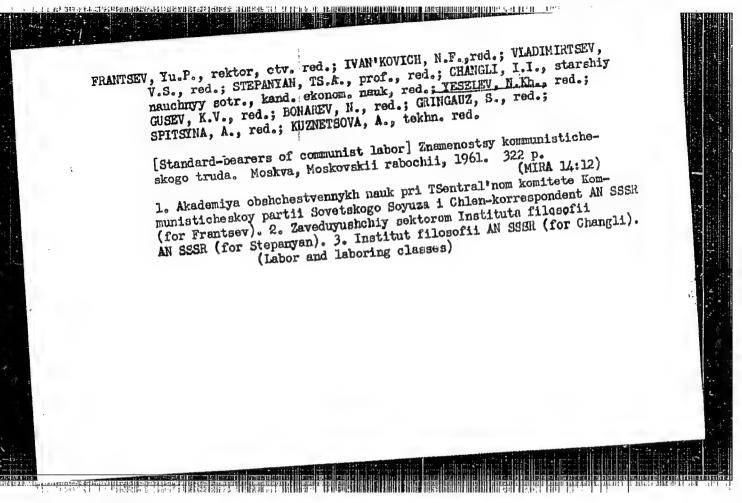
Comparative diagnostic value of the iodine test and aminopherase activity in myocardial infarction; an abstract. Lab. delo. no.1:

25 '65.

1. Kafedra gospital'noy terapii (zaveduyushchiy - prof. L.S.

Shvarts) lechebnogo fakul'teta Saratovskogo meditsinskogo inShvarts) lechebnogo fakul'teta Saratovskogo meditsinskogo inShvarts i l'ya gorodskaya klinicheskaya bol'nitsa im. V.I. Lenina (glavnyy vrach Yu.Ya. Gordeyev).





YESELEVICH, A. YA.

21035 Yeselevich, A. Ya., Shiryak, E.A. i i istovskaya, I.M. Lecheniya infitsirovannykh
Ran chudesnoy palochkoy Trudy In-ta (Kazansk Nauch-issled In-t ortopečii i vosstanovit Kirurgi
t.111, 1949, s. 206-19,

SO: LETOPIS ZHURNAL STATEY - Vol. 28, Moskva, 1949

YESELEVICH, A. YA.

21048 Spasskiy, N.N. i Eselevich, A. Ya. Issledovaniya Kul'Tural'No-F:rmentativnykh i patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh pri Khronicheskih infektsiyakh Ran Trudy patogennykh svoystv stafilokokov, Vydelennykh svoystv stafilokokov, Vydelennykh svoystv stafilokokov, Vydelennykh

### YESELEVICH, A.Ya.

Characteristics of hyaluronidase-active staphylococci isolated from chronic wound processes in veterans of Warld War II. Ortop.travm. i protes. no.5:48-52 S-0 155. (HLRA 9:12)

1. Iz Kazanskogo nauchno-issledovatel'skogo instituta ortopedii i vosstanovitel'noy khirurgii (dir. - zasluzhennyy deyatel' nauki TASSE prof. L.I.Shulutko)

(MICROCOCCUS PYOOMNES
isolated from chronic war wds., hyaluronidase activity)
(WOUNDS AND INJURIES.
war wds., isolation of Micrococcus pyogenes, hyaluronidase activity)

YESRIEVICH, A. YA.

Yeselevich, A. Ye.

"Hyaluronidase of Staphyloccoci Isolated in Cases of Chronic Wound Processes." Kazan' Sci. Res Inst of Orthopedics and Restorative Surgery. Kazan', 1955 (Discertation for the degree of Candidate in Medical Science)

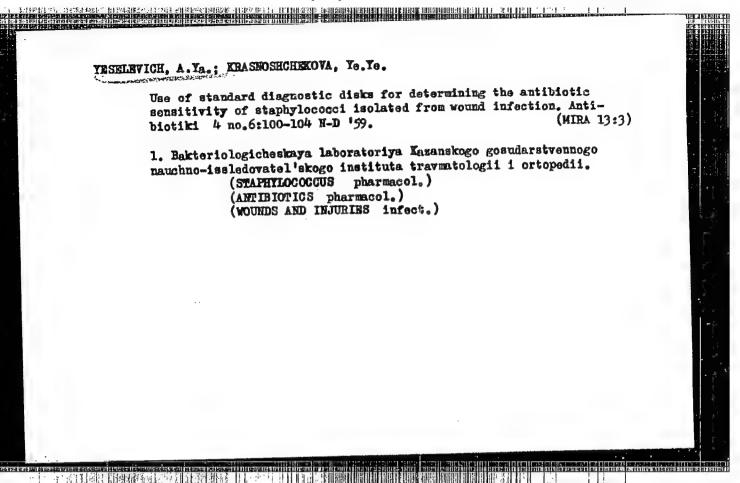
SO: Knizhnaya letoris' No. 27, 2 July 1955

YESRLEVICH, A.Ya.; PODVAL'NYY, A.Yu.

Microflore of fresh wounds of peacetime injuries. Ortop.travm. i
protes. 17 no.6:122-123 H-D '56.

1. Iz Kasanskogo nauchno-issledovatel'skogo instituta vosstanovitel'noy khirurgii i ortopedii (direktor - zasluzhennyy deyatel' nauki TASSR
professor L.I.Shulutko)

(VOUNDS--RACTERIOLOGY)



YESELEVICH, A. Ya.; KRASMOCHCHEMOVA, Ye.Ye.

Characteristics of penicillin-resistant staphylocodd isolated from wound infections. Antibiotiki 8 no. 52478-279 (MIRA 17:2)

1. Kazanskiy nauchab-issledovatel skiy institut travmatologii i ortopedii.

ABDRASHITOVA, L.S.; YESELEVICH, A.Ya.; KRASNOSHCHEKOVA, Ye.Ye.

Microflora in children with odontogenic osteomyelitis. Stomatologiia 42 no.4840-42 Jl-Ag'63 (MIRA 17:4)

1. Iz Tatarskoy respublikanskoy stomatologichoskoy bol'nitsy (glavnyy vrach S.Z. Zalyayutdinova) i bakteriologicheskoy laboratorii Kazanskogo gosudarstvennogo nauchno-issledovateli skogo instituta travmatologii i ortopedii (dir. - kand. med. nauk U.Ya. Bogdanovich).

VESE/Human and Animal Physiology (Normal and Pathological). T-12

Nervous System. Epilepsy.

Abs Jour : Ref Zhur - Biol:, No 11, 1958, 51264

Author : Yeselevich, E.I.

Inst : Chkalov Institute of Medicine.

Title : Clinical and Experimental Data on the Significance of the Corpus Callosum in the Structure of Epileptic Seizures.

Orig Pub : Tr. Chkalovsk. med. in-ta, 1956, vyp. 5, 420-426.

Abstract: After the corpus callosum was severed in rats, no essential changes of epilepsy produced by ringing sounds were observed. The only finding was that in operated rats the cataleptic stage of the seizure appeared sooner and lasted somewhat longer than in healthy animals. Apparently, the corpus callosum does not play an important role in the struc-

ture of spasmotic seizures in cpilepsy which was induced

Card 1/2

- 95 -

YESELEVICH, E. 1.

USSR / Pharmacology, Toxicology. Toxicology.

V

Abs Jour: Ref Zhur-Biol., No 9, 1958, 42520.

Author : Yeselerich B. I

Inst : Chkalov Medical Institute.

Title : On the Symptomatology of DDT poisoning.

Orig Pub: Tr. Chkalovskogo med. in-ta, 1956, vyp. 5, 427-429.

Abstract: DDT powder was used by mistake, instead of flour, in the preparation of butter fried potato patties. Ingestion of 15-20 gm of DDT dust caused not-lethal poisoning of children and adults. It is possible that the toxicity of DDT was decreased by heating, by the acidity of the products, by fermentation in the dough etc. Rapidly following manifestations: stomach ache, headache, intoxication of the CNS. The symptoms of CNS damage in children-inhibition or excitation of the subcortical motor mechanisms, in adults- disorders of corrdination of movements.

Card 1/1

67

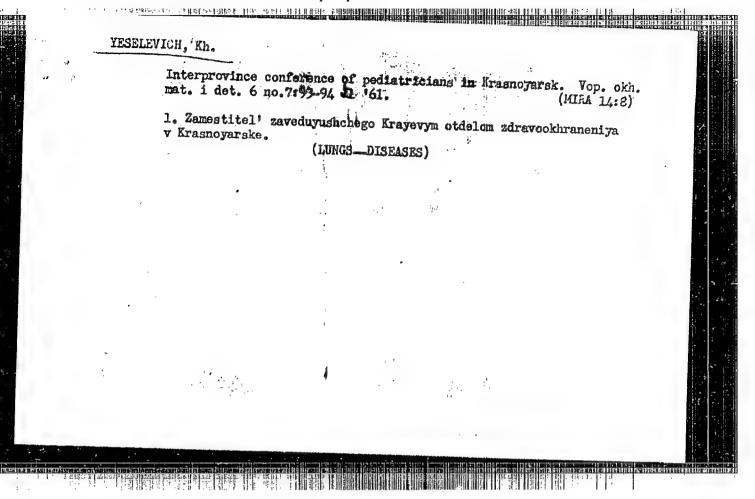
YESELEVICH, E. I. (Prof.) Chkelov

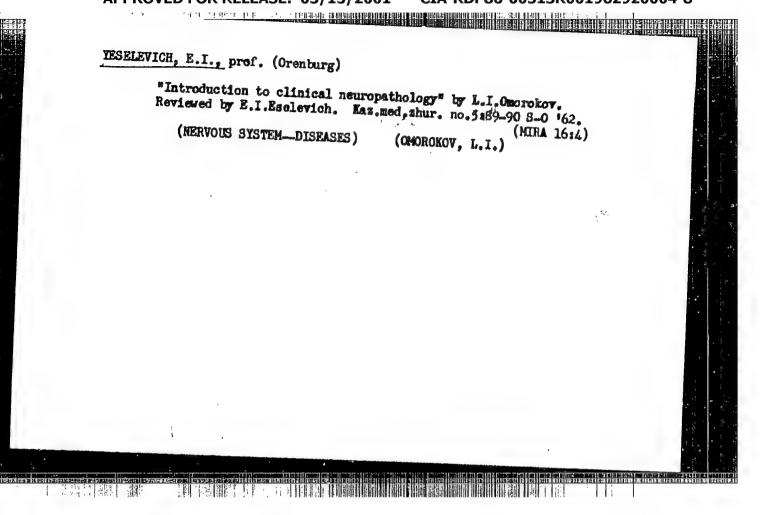
3. M.

Klinika i Lecheniye Newmo-Psikhicheekikh Nerusheniy pri Ku-likoredke (with Abramenko, Poliomielit v Chkelovskoy Oblasti B 1946-1955gg.

A. I. and others p. 413 V sb Aktual'nyy Problemy Newropatologii i Psikhiatrii. Kuybyshev. 1957.

Iz kafedry nervnykh bolezney Chkelovskogo gosudarstvennogo meditsinskogo instituta, Zaveduyushchiy kafedroy - Prof. E. I. Yeselevich.





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18(5)

507/132-59-4-1/17

AUTHORS:

Weselevich L.V., Lisitsyr, A.I., Luchiz, N.S.

TITLE:

The Ancient Zircon-Ilmenite Placer in the Mesc-Cenozoic Deposits of West Siberia.

PERIODICAL:

Razvedka i okhrana nedr, 1959, Nr 4, pp 1-4

ABSTRACT: The m

The Tuganskoye zircon-ilmenite placer was discovered in 1956-1957. It is located on the water divide of the rivers Tom and Yaya in the region of northern spurs of the Kuznetskiy Alatau mountain range. The Paleozoic foundation of metamorphic rocks of the region is covered by an erosion crust, 15 to 70 m thick, formed under continental conditions during a period from the Middle-Carbonical ferous up to Upper-Gretaceous and even Paleogene times. This crust covers both slopes of the water divide of the rivers Tom and Yaya. Zircon and

Card 1/3

### CIA-RDP86-00513R001962920004-8 "APPROVED FOR RELEASE: 03/15/2001

307/132-59-4-1/17

The Ancient Zircon-Ilmenite Placer in the Meso-Genozdic Deposits

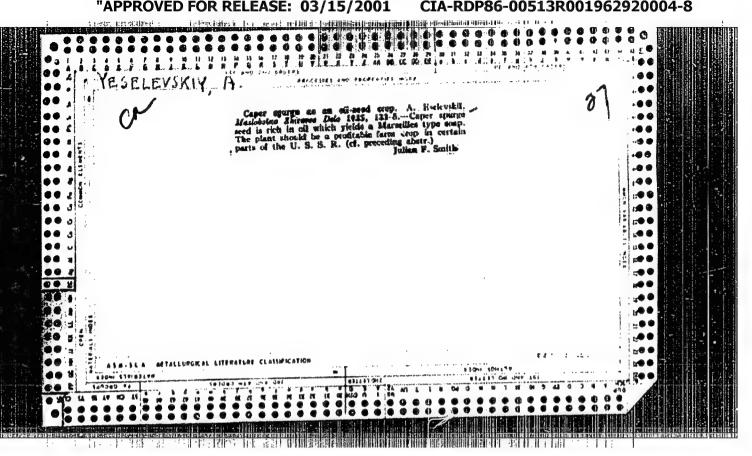
ilmenite were found in this stratum formed by the metamorphic rocks and the erosion crust. In Paleoffene time, this weathered crust was again eroded by the transgressing sea, the clay fraction was washed away in the sea and the coarse-grained fraction was deposited in the coastal area. These deposits at present are divided into three suites, by their granulometric composition, the Simonovskaya, the Mariinskaya and Tuganskaya suites. The rare elements are found mainly in the Tuganskaya Tuite composed of variously grained sands. Consistenal selective concentrates can be obtained from these sands. The Tugans oye deposit can be emploited by opencast mining.

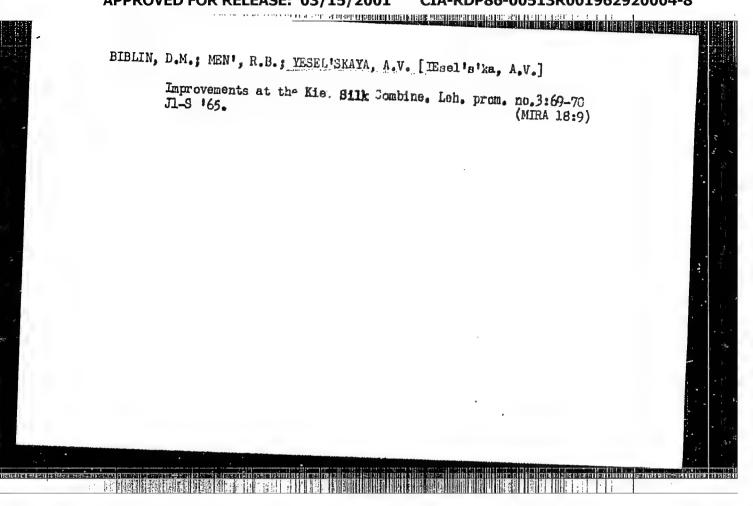
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CIA-RDP86-00513R001962920004-8" APPROVED FOR RELEASE: 03/15/2001

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SOV/132-59-4-1/17 The Ancient Zircon-Ilmenite Placer in the Meso-Cenozoic Deposits of West Siberia. ASSOCIATION: Ministerstvo geologii i okhrany nedr 335R. (The Ninister of Geology and Conservation of Mineral Resources of the USSR. (Yeselevich, Lisitsyn, 12, 2) Giredmet (Pyatnov) Card 3/3

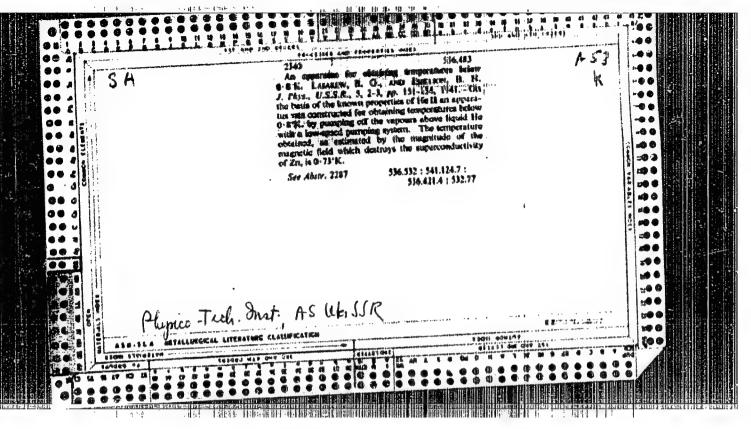


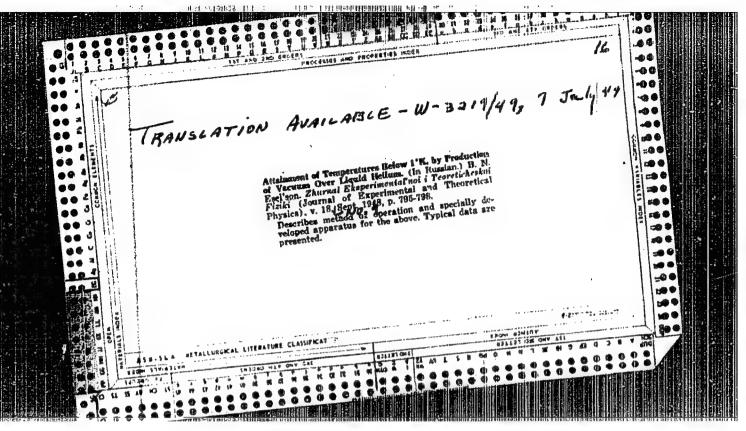


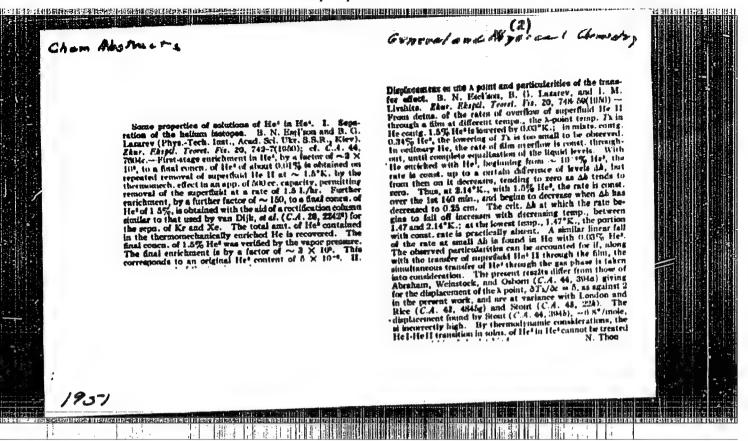
TERENT'YEV, A.P.; VIKTOROVA, Ye.A.; YESEL'SON, B.M.; KOST, A.N.; YERSHEV, V.V.

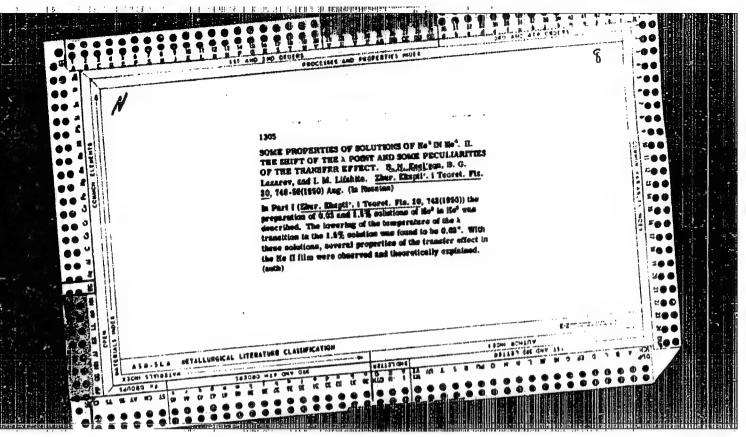
Inner complex compounds as contact insecticides. Zhur.ob. khim. 30 no.7:2422-2427 Jl '60. (MIRA 13:7)

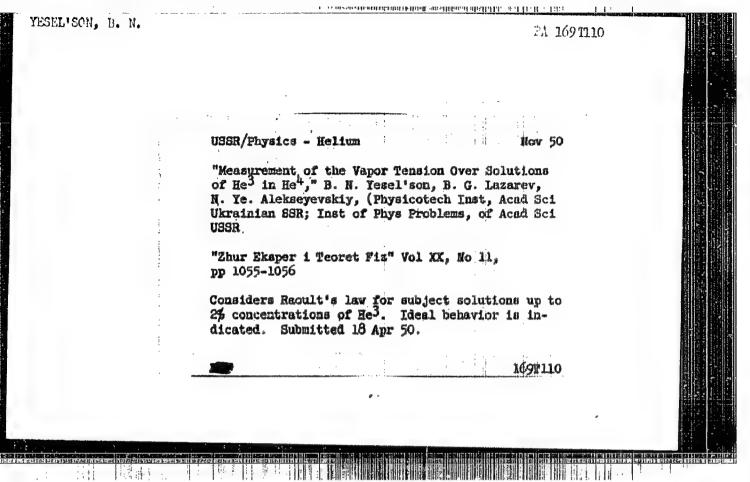
1. Moskovskiy gosudarstvennyy universitet.
(Complex compounds) (Insecticides)

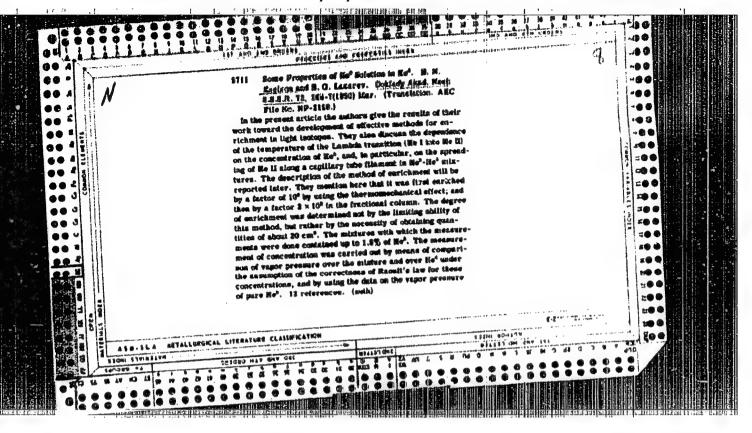


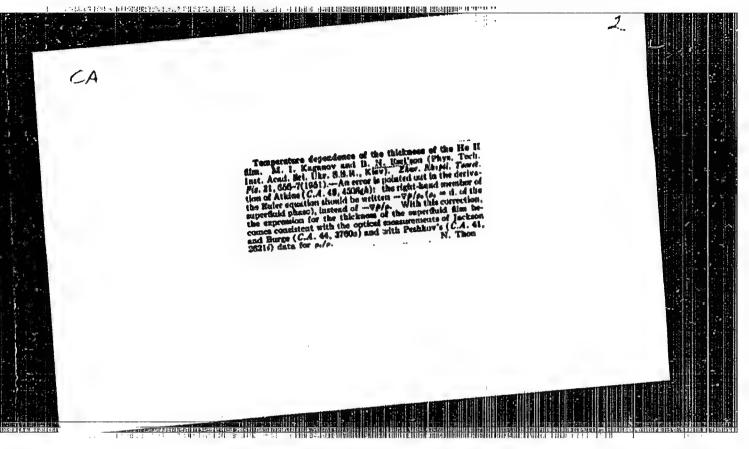


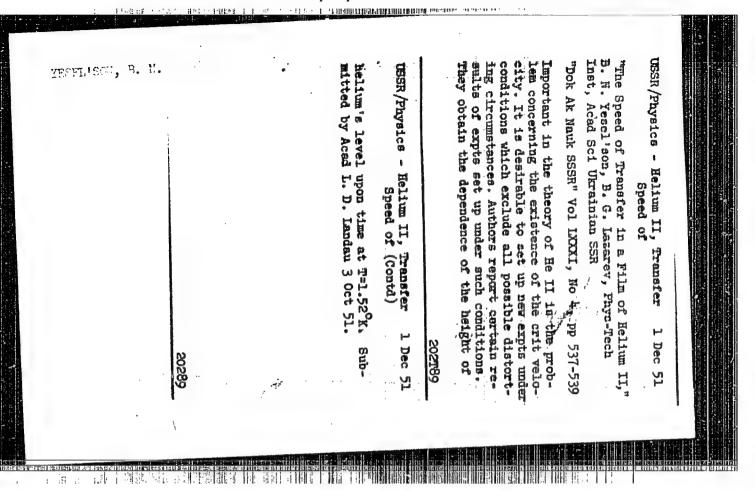


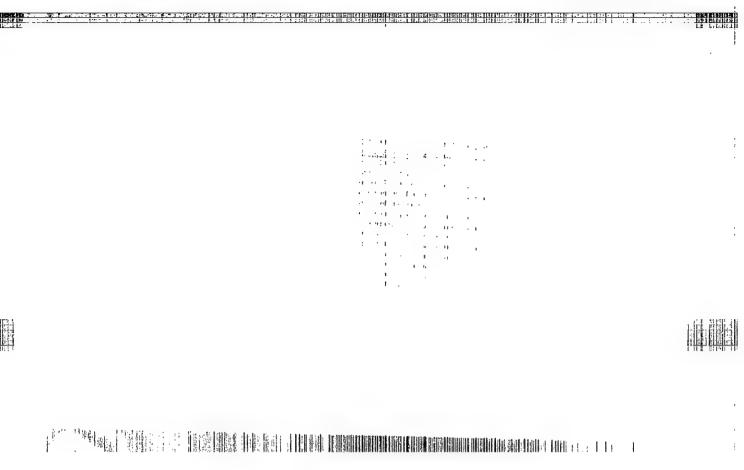












USSR/Physics - Liquid Helium He-3

FD-722

Card 1/1

: Pub 146-10/18

Author .

: Yesel'son, B. N.

reser son, b. n.

Title

: Some properties solutions of He3 in He4. III. Vapor tension

Periodical

: Zhur. eskp. i teor. fiz., 26, 744-750, Jun 1954

Abstract

: The vapor tension of solutions of He<sup>3</sup> in He<sup>4</sup> is measured under conditions ensuring the establishment of equilibrium between the liquid and the vapor. Data obtained from solutions containing He<sup>3</sup> in the amounts of 0.49, 1.00, 2.40, 4.23, 5.18 and 8.08% allow one to conclude that the behavior of such solutions is not ideal. Indebted to B. G. Lazarev, N. Ye. Alekseyevskiy, I. M. Lifshits, Ye. S. Borovnik.

14 references, including 10 foreign.

Institution

: Physicotechnical Institute, Acad. Sci. USSR

Submitted

: October 10, 1953

### "APPROVED FOR RELEASE: 03/15/2001 CIA-RDP86-00513R001962920004-8 TO THE PROPERTY OF THE PROPERT

USSR/Physics - Helium isotopes

FD-991

Card 1/1

Pub. 146 - 15/20

Author

: Yesel'son, B. N., and Bereznyak, N. G.

Title

: Dew points of mixtures of helium isotopes

Periodical

: Zhur. eksp. i teor. fiz., 27, No 5 (11), 648, 649, Nov 1954

Abstract

: The authors tabulate the dependence of the pressure of initial condensation upon temperature for mixtures with various contents of helium-3, and graphs the dependence of the vapor tension of mixtures of helium isotopes upon the state of the gaseous phase for various temperatures. Such tabulation and graphing are necessary in order for the authors to construct the vapor-liquid diagrams for the system He3 He4. An extension of an earlier work (B. N. Yesel'son, ibid., 26, 744, 1954). A detailed report will be published soon. The authors thank professor N. Ye. Alekseyevskiy for analyzing the mixtures for the content of the light isotope and pro-

fessor B. G. Lazarev for his interest.

Institution : Physicotechnical Institute, Academy of Sciences Ukrainian SSR

Submitted

: July 13, 1954

CIA-RDP86-00513R001962920004-8" APPROVED FOR RELEASE: 03/15/2001

USSR/ Physics Isotopes

Card : 1/1

Authors : Eselison B. N. and Lazarev, B. G., Act. Memb. of Acad. of Sc.

TO SERVICE CONTRACTOR OF THE SECOND CONTRACTOR

Ukr-SSR

Fitte : Solidification of helium isotope mixtures

Feriodical : Dokl. AN SSSR, 97, Ed. 1, 61 - 64, July 1954

Abstract: Data are presented on the solidification point of pure Heli as well as helium isotope mixtures obtained by a previously described method.

The experimental installation and the investigation procedure are described. The data obtained (shown in graph) make it issuing to evaluate the nature of the structural diagram for inquit and solid phases of the Head - Head system. The pressure at which telium solidifies was recorded with greater accuracy by means of two manemeters the indications of which coincided with each other only as long as the helium remained in liquid state. Mine references:

USSR, 4 USA, 2 German. Graphs, drawing.

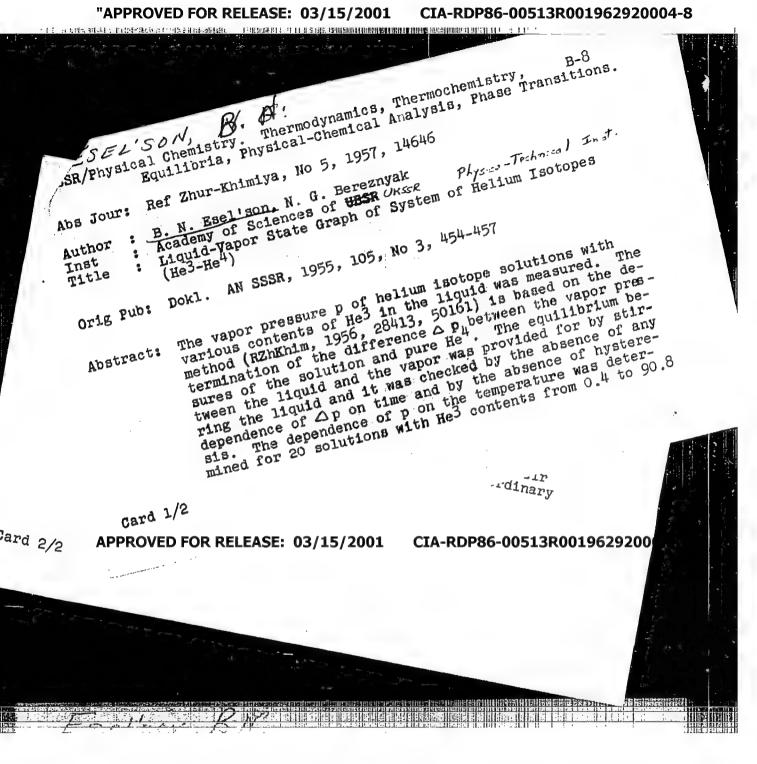
Institution: Acad. of Sc. Ukr-SSR. Physico-Technical Institute

Submitted : Haron 25, 1954

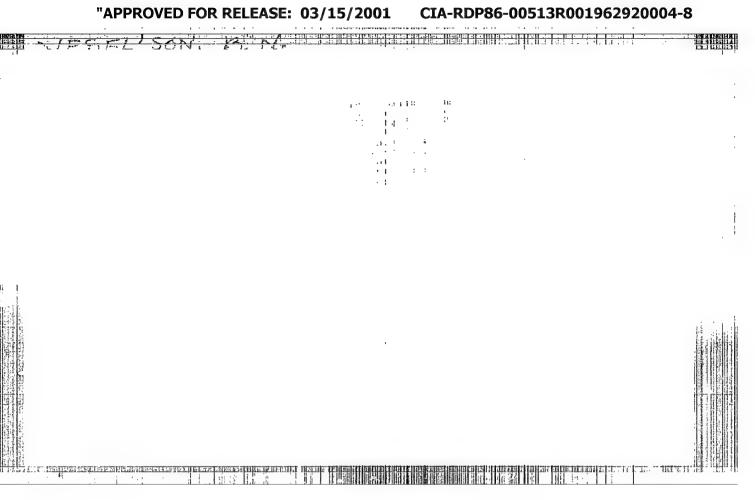
SECTION OF THE SECTIO YESEL SON, B.N. USSR/Physics - Surface tension 1 Pub. 22 - 15/49 Card 1/1 : Esel'son, B. N., and Bereznyak, N. G. Authors and the first the first of the f Surface tension of helium isotope solutions Title Periodical # Dok. AN SSSR 98/4, 569-571, Oct. 1, 1954 an experiment was conducted with solutions of helium isotopas to Abstract determine their surface tensions. The mothod and instrument set-1: are outlined. Six references (1921-1944). Diagram: graphs. Institution : Physico-Technical Institute of the Acad. of Scs. of the Ukr. SSR Presented by : Academician Lindau, L. D., April 22, 1954

YESEL SON, B.N. USSR/Physics - Surface tension Card 1/1 Pub. 22 - 7/40 s Esel'son, B.N., and Bereznyak, N.G. Authors \* Surface tension of a light helium isotope Title Periodical : Dok. AN SSSR 99/3, 365-367, Nov 21, 1954 The experimental determination of the surface tension of a light helium Abstract isotope(He3) is described. The following formula was used for this deterinto which the experimental data mination:  $2\alpha(\vec{b}_1 - \vec{b}_2) = H_g(\vec{g} + \vec{g}_{e'})$  obtained was substituted. Symbols are explained. Five references: 1-USSR (1921-1954). Diagram, table; graph. Physico-Technical Institute of the Acad, of Scs. of the UkrSSR. Institution: Presented by: Academician L.D. Lindau, July 12, 1954

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YESEL'SUN, BN.

SUBJECT USSR / PHYSICS CARD 1 / 2 PA - 1779
AUTHOR ESEL'SON, V.H., LAZAREV, B.G., SINEL'NIKOV, K.D., SVEC, A.B.

AUTHOR ESEL'SON, V.R., LAZAREV, B.G., SINGE TITLE On Some Peculiarities of Rotating He II.

TITLE On Some Peculiarities of Rossian 2012 (1956)
PERIODICAL Zurn.eksp.i teor.fis, 31, fasc.5, 912-912 (1956)

Issued: 1 / 1957

At first several previous works dealing with this topic are cited. An experimental confirmation of the dependence of the inertia moment of rotating He II on velocity and an estimation of relaxation time would be most desirable. This problem could be solved by studying the damping of the rotation of a glass with He II which is the nearest approach to the continuous equilibrium between the normal and the supraconductive component. As relaxation time was not known, the rotating system had to have asufficiently low damping. For this purpose a plexiglass vessel was suspended in a magnetic field which warranted rotation of the vessel for several hours after an initial velocity of several revolutions per second had been imparted to it. The vessel (R = 1,5 cm) contained about 300 light aluminium disks which were arranged at a shorter distance than the depth of penetration of the viscous wave. With the help of a rotating magnetic field the rotation velocity of the vessel containing the He II was brought up to the assumed value, after which the field was switched off. Under these conditions only the normal component of the He II could at first be taken away with the disks, but with its supraliquid component this was possible only after relaxation time. If relaxation time exceeds the time of screwing-out (?), it was obvious that, with a growing distance of the supraliquid component, a consider-

全国企业中的企业,企业中的企业的企业,但是一个企业,以上的企业,以上的企业,以上的企业的企业,但是一个企业中的企业,但是一个企业的企业,但是一个企业的企业,但是 1985年中国企业的企业,在1985年的企业,1985年的企业,以上的企业的企业的企业,以上的企业的企业,以上的企业的企业的企业,以上的企业的企业的企业的企业的企业,以上的企业的企业的企业的企业的企业。

Zurn.eksp.i teor.fis,31,fasc.5,912-912 (1956) CARD 2 / 2

PA - 1779

able modification of the moment of inertia of the vessel containing the helium (about 25%) was to be expected, which would mean a modification of rotation velocity.

However, the investigation of the damping of the rotating vessel containing the He II showed no noticeable change of velocity, which is illustrated by an attached diagram for the dependence of rotation velocity on time recorded at  $T=1,5^{\circ}$  K for a duration of screwing out (?) of 10, and for 2 seconds. The same tests make it possible to determine the dependence of the inertial moment of He II on rotation velocity. It was found that at velocities of more than 0,5 rotation per second there is no such dependence.

Thus, the lack of the extraction of the supraconductive component on the occasion of experiments with an oscillating stack of disks when small amplitudes are used can by no means be explained by too long a relaxation time. Hitherto, the problem of the dependence of relaxation time on velocity has not yet been solved. The authors' attention was drawn to this fact by L.D.LANDAU.

INSTITUTION: Physical-Technical Institute of the Academy of Science of the Ukrainian SSR.

YESEL SON, B.N.

USSR/Nuclear Physics

C-5

Abs Jour

: Referat Zhur - Fizika, No 5, 1957, 11238

Author

: Klyucharev, A.P., Yesel'son, B.N., Val'ter, A.K.

Inst

: Physical-Technical Institute, Academy of Sciences.

Ukrainian SSR

Title

: Study of the Reaction Between He3 and Deuterons.

Orig Pub

: Dokl. AN SSSR, 1956, 109, No 4, 737-739

Abstract

: The excitation function of the He<sup>3</sup> (d, p) He<sup>4</sup> was measured in the deuteron energy regions up to 1.5 Mev. The deuterons were accelerated by an electrostatic generator, the energy scale of which was calibrated by the resonant maxima of the excitation function of the F<sup>19</sup> (p, × )0<sup>16</sup> reaction. A gas target with a window of aluminum foil approximately five microns thick was used, filled with helium at a pressure of 50 rm mercury, and containing 57.6% He<sup>3</sup>.

Card 1/2

USSR/Nuclear Physics

C-5

Abs Jour : Ref Zhur - Fizika, No 5, 1957, 11238

To obtain energies below 450 kev, the deuterons were slowed down by aluminum foils. The overall error in the determination of the deuteron energy amounted to I 30 kev in the region of the resonance of the excitation curve. The particles were registered at an angle of 90° to the direction of the deuteron beam by means of a proportional counter with a mica window. The excitation function obtained has a resonant maximum at a deuteron energy of 435 kev, corresponding to the formation of an intermediate Li<sup>5</sup> nucleus in a state having an excitation energy of 16.8 Mev. The absolute cross section in the resonance is 63.4 % 3.2 millibarns per steradian.

Card 2/2

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SUBJECT USSR / PHYSICS CARD 1 / 2 PA - 1978

AUTHOR BEREZNJAK, N.G., ESEL'SON, B.N.

TITLE The Energy Spectrum of He-3 Admixtures dissolved in He II.

TITLE The Energy Spectrum of hear Admitsuated and PERIODICAL Dokl. Akad. Nauk 111, fasc. 2, 322-324 (1956)

Issued: 1 / 1957

An experimental investigation of the temperature dependence of the contribution of the admixtures to the density of the normal He II component permits n ad a univocal determination of the shape of the energy spectrum. For this purpose, a univocal determination of the shape of the energy spectrum. For this purpose, the authors measured the density of the normal component of the solution of He? in  $\text{He}^4$  with a concentration of x = 3,0%  $\text{He}^3$ . The temperature dependence of the moment of inertia of a stack of light parallel disks steeped into the heliumisotope solution was measured. The stack of disks was firmly connected to the little pail surrounding it. The latter was suspended on a wire of phosphorous bronze so that it could perform rotating oscillations round an axis which was vertical to the plane of the disk. The modification of the moment of inertia of the device was determined from the temperaturedependence of the period of the oscillations of the system in the liquid. The connection between the oscillation period of the system and the liquid participating in the motion of the device can, as usual, be determined by solving the corresponding hydrodynamic problem. It must, however, be considered that the liquid is drawn off not only by the disks but also by the outer surfaces of the pail. When solving the hydrodynamic problem the peculiarities of the experimental device must be taken into account by imposing certain corresponding boundary conditions. In this way two equations

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PA - 1978 Dokl.Akad.Nauk 111, fasc. 2, 322-324 (1956) are obtained of which one permits determining the penetration depth  $\delta$  and the other the determination of the density of the normal component. Both equations are explicitly given. By means of the device described the temperature dependence of the density of the normal component of pure He4 and of a solution of helium isotopes with a content of 3.0% He3 was determined. The results are shown in form of a diagram and are indicative of the fact that the normal component of the solution has a considerably greater density than He<sup>4</sup>. This follows also from the theory by I.JK.POMERANČUK. At 1,5°, Qn/Q is by 50% greater in the case of the solution than with He4. The spectrum of elementary excitations which corresponds to the particles of the admixture is characterized by the value p = 0. (Here p apparently denotes the pulse in the case of a lacking admixture). From the experimentally determined values of  $(Q_n/Q_{\lambda})_g$  for the solution and  $(Q_n/Q_{\lambda})_g$ for pure He4 it is possible to determine the effective mass of the admixture in the solution. Such a computation furnishes the value  $\mu = 2,5 \text{ m}_2$ , where m<sub>2</sub> denotes the mass of the He3-atom. At present experiments for the determination in concentrated mixtures are being carried out. of  $Q_n/Q_{\lambda}$ 

INSTITUTION: Physical-Technical Institute of the Academy of Science in the Ukrainian SSR.

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Dokl. Akad. Nauk 111, fasc. 3, 568-570 (1956) CARD 2 / 2 In general the determination of such a break on the curve P(T) is difficult, but it is considerably facilitated by the study of the temperature dependence of the difference  $\Delta$  P of the vapor pressure of the solutions and of pure He4. In the case of the curve P-T the relatively small discontinuity of this quantity at the A-point will be only little noticeable. However, in the case of the curve  $\triangle$  P-T the value of  $d/dT(\triangle$  P) diminishes considerably and the discontinuity of this quantity at the  $\lambda$ -point remains the same. A diagram illustrates the dependence P-T for some solutions. In the case of all these curves which were obtained by the differential method of measuring vapor pressure a discontinuity is observed which must currespond to the temperature of the phase transition. These temperatures and the corresponding concentrations of the solutions are shown together in a table. These data deviate considerably from the results obtained by other works. However, the data found here agree well with those values of Ty which were obtained recently in connection with the study of various properties of the solutions of He in He within the domain of small concentrations. The value of  $(dT_{\lambda}/dx_{fl})$ at x = 0 can be obtained by using the data concerning the density of the normal component of the solutions of helium isotopes. The here computed value of  $(dT_{\lambda}/dx_{fl})$  at  $x_{fl} = 0$  agrees well with the values -1,5  $\nabla$ /mol which were INSTITUTION: Physical-Technical Institute of the Academy of Science in the

YESEL'SON, Boris Naumovich (Physico-Tech Sci-Res Inst, AS, UKSSR) awarded sci degree of Doc Physico-Math Sci for the 21 Jun 57 defense of dissertation: "Research on the properties of helium isotopes and solutions of them" at the Council, Khar'kov State Univ imeni Gor'kiy; P rot No 11, 10 May 58.

(BMVO, 10-58,20)

ALTER GENERAL MARKET MARKET BELLEVILLE STATE OF THE STATE PESEL SON, BN 56-4-18/54 Kaganov, M.I., Lifshits, I.M. Yesel'son, B.N., The Thermodynamics of the Phase Transition between He I and AUTHORS: He II in Solutions of Helium Isotopes (Termodinamika fazovogo perekhoda He I - He II v rastvorakh izotopov geliya) TITLE: Zhurnal Eksperim. i Teoret. Fiziki, 1957, Vol. 33, Nr 4, pp. 936 - 944 (USSR) PERIODICAL: 1.) The phenomena that are connected with the transition from He I and He II in solutions of helium isotopes are thermodynamically (theoretically) treated. It is shown that this transition, in the range from 1,35 to 3,0 K, is a second type phase ABSTRACT: 2.) It is shown that at the temperature of the second type phase transition a point of sudden irregulatity should occur in the derivative as well of the partial as of the total pressure according to the temperature, which fact is experimentally in dependence on the distribu-3.) It is shown that at The in dependence on the distribution coefficient, a point of sudden irregularity should be observed in the derivative according to the temperature. a point of sudden irregularity 4.) It is shown that at TA Card 1/2

The Thermodynamics of the Phase Transition between He I and He II in Solutions of Helium Isotopes

should be observed for the heat of solution and the heat of vaporization. For weak solutions numerical data are given for the point of sudden irregularity of the heat of solution. For one solution the course of curve of the heat of solution is also calculated. There are 6 figures and 7 Slavic references.

ASSOCIATION:

Physico-Technical Institute AN Ukrainian SSR

(Fiziko-tekhnichseskiy institut Akademii nauk Ukrainskoy SSR)

SUBMITTED:

April 19, 1957

AVAILABLE:

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SOV/81-59-24-84747

Translation from: Referativnyy zhurnal. Khimiya, 1959, Nr 24, p 9 (USSR)

Klyucharev, A.P., Val'ter, A.K., Yesel'son, B.N. AUTHORS:

The Reaction of He3 With Deuterons TITLE:

Tr. Sessii AS UkrSSR po mirn. ispol zovaniyu atomn. energii. Kiyev, AS PERIODICAL:

Ukrssr, 1958, pp 64 - 69

The measurement of the differential cross section of the reaction He3 ABSTRACT:

(d, p) He at deuteron energies of 100 - 1,500 kev is reported. & -particles were recorded which escaped under an angle of 90° to the direction of the deuteron beam. The dependence of the cross section on the energy has a resonance course with a maximum at  $E_{\rm d} \approx 435$  kev. The value of the cross

section at the maximum is 63.4 mbarn-sterad.

V.R.

Card 1/1

XESAL'SON IS in.

AUTHORS: Yesel'son, B. N., Shvets, A. D., Bablidge, R. A. 56-1-38/56

TITLE: On the Film Flow Rate in Solutions of Helium-Isotopes

(O skorosti perenosa po plenke u rastvorov izotopov geliya)

PERIODICAL: Zhurnal Eksperimental noy i Teoreticheskoy Fiziki, 1958,

Vol. 34, Nr 1, pp. 233-234 (USSR)

ABSTRACT: The influence of He<sup>3</sup> dissolved in He II upon the film flow is, as is well-known, reduced to the decrease in the flow

rate. It was of a certain interest to investigate this fact more thoroughly and therefore the authors made tests with a solution of helium-isotopes with a helium content of 1,5; 4,7; 7,0 and 9,6%. The apparatus used for these tests consists of two elbows of a thin-walled capillary tube (diameter 1,08 mm) of equal lengths communicating over a helium-film. The film flow rate  $R = v\delta$  was measured by the measurement of the rate of change of the liquid level in

one of these elbows. In this connection v signifies the rate of the motion of the film and  $\delta$  - the thickness of the film. The temperature interval immediately following the  $\lambda$  -

-point was investigated. The results obtained here are

Card 1/3 illustrated in two diagrams. One of these diagrams

On the Film Flow Rate in Solutions of Helium-Isotopes

56-1-38/56

illustrates the dependence of the film flow rate on the temperature and the other diagram — the dependence of the film flow rate on the content of He. According to the results found here the film flow rate increases with increasing concentration of He. When having data on the dependence of the density on the temperature for the solutions of the helium-isotopes, the following conclusions can be drawn: The film flow rate in the temperature range investigated here is directly proportional to the density of the superliquid component:  $R = Ac_g/e$ , where  $A = 3,2.10^{-5}$  cm/cmsec. Moreover the temperature of the phase transition He I — He II for the solutions given here might be determined from the beginning of overflowing over the film. The values obtained in this connection are in satisfactory agreement with the analogous results obtained by other methods. There are 2 figures, 2 tables, and 5 references, 4 of which are Slavic.

ASSOCIATION:

Physical-Technical Institute AN Ukrainian SSR (Fiziko-tekhnicheskiy institut Akademii nauk Ukrainskoy SSR)

Card 2/3

The Use of a Superconductive Ring for Recording the SOV/56-37-1-61/64 Phase Transition in Liquid Helium

mental results are shown by a diagram. It shows the deviation  $\alpha$  of the mirror galvanometer connected to the immobile coil in dependence on the time t (during which helium temperature increases). The curve  $\alpha(t)$  has a slight  $\alpha$ -decrease with an increase of helium temperature from 1.5 K to the  $\lambda$ -point (from t=0 to t=17.5 min), after which there is a sharp increase (to about 30 times its amount) in the  $\lambda$ -point, corresponding to the jump of thermal conductivity at this point. This narrow and steep peak is followed by a second small maximum; At T = 3.73 K and t = 36.5 min,  $\alpha$  decreases to zero. In these experiments helium was heated by light; in the case of electrical heating, the peak in the  $\lambda$ -point of the  $\alpha(t)$ -curve does not discussing the results. There are 1 figure and 2 Soviet references.

ASSOCIATION:

Fiziko-tekhnicheskiy institut Akademii nauk Ukrainskoy SSR (Physico-technical Institute of the Academy of Sciences,

SUBMITTED: Card 2/2

May 13, 1959

24(8)
AUTHORS: Yesel'son, B. N., Kaganov, M. I., Lifshits, I. M.

TITLE: Reply to the Letter by M. P. Mokhnatkin (Otvet na pis'mo

M. P. Mokhnatkina)

PERIODICAL: Zhurnal eksperimental'noy i teoreticheskoy fiziki, 1959,

Vol 36, Nr 3, p 964 (USSR)

ABSTRACT: In a "Letter to the Editor", Mokhnatkin criticized a paper

by the authors of this "reply", and declared that terms were omitted in two formulae. In this reply these omissions are described as being justified, and it is pointed out that in all cases in which it was found necessary, these terms were

mentioned. In this connection a formula is specifically

mentioned.

SUBMITTED: November 25, 1958

Card 1/1

33155 5/120/61/000/006/026/041 E032/E114 24.5600 Yesel'son, B.N., Shvets, A.D., and Bereznyak, N.G. AUTHORS: An He 3 apparatus for the production of temperatures TITLE: down to 0.3 oK PERIODICAL: Pribory i tekhnika eksperimenta, no.6, 1961, 123-124 The apparatus is illustrated in the figure. About 2 litres of gaseous He3 supplied by the cylinders 1 are condensed into the copper container 2 which is located inside the vacuum envelope 3 and is maintained at the temperature of the outer bath (1.3 °K). Since at this temperature the vapour pressure of He3 is greater than the pressure at which diffusion pumps begin to operate, there is an additional He bath 4 whose temperature may be reduced to 1 °K by pumping the vapour through a diaphragm by the APH-50 (DRN-50) pump 5. The valve 6 is used to fill this bath with liquid He<sup>4</sup> from a dewar. Under these conditions the vapour given off by liquid He3 may be pumped by the mercury diffusion pump (Leybold) 7 which has a pumping speed of about 15 litres/sec. Mercury vapour is excluded by liquid nitrogen traps. The He3 vapour pumped by 7 is Card 1/ # 4

33155

An He 3 apparatus for the production...

S/120/61/000/006/026/041 E032/E114

continuously removed by the liquid-hydrogen cooled charcoal pump 8 containing about 50g of activated charcoal. In this way the He3 gas can be recovered and returned into the reservoirs 1. The use of these absorption pumps greatly simplifies the design of cryostats containing Ha?. It was found, convenient to use a solution of He3 in He2 instead of pure He2 as the cooling medium. To do this, a mixture containing 7.4% of He3 was condensed through the tube 9 into the glass reservoir 10 which was through the tube y into the glass reservoir to missing sealed into the He3 container through a Kovar seal. Since this cryostat was used to study the properties of He3 + He4 mixtures, the reservoir 10 contained the glass vessel 11 which was filled with the mixture under investigation through the tube 12, It was found that the minimum temperature was 0.4 oK and could be maintained for about 6 hours, which is much longer than the period obtained with He as the cooling liquid. The lower temperature of 0.3 °K was obtained by pumping the vapour given off by liquid He3 placed in a very small glass dewar connected to the pumping system described above. The latter temperature could be maintained for over 7 hours. Temperatures between Card 2/# (/

s/120/61/000/006/026/041 33155 An Ha 3 apparatus for the production . E032/E114 1 and 0.4 °K, could be obtained by adjusting the pumping speed of the diffusion pump with the aid of the valve 13. In all the experiments the temperature was determined by measuring the He3 vapour pressure with a McLeod gauge (Ref, 14; S.G. Sydoriak, T.R. Roberts, Phys. Rev., v. 106, 1957, 175). In one of the experiments the He3 vapour was pumped by the absorption pump only the pump being cooled by liquid helium (4.2 °K). In spite of the long and narrow connecting pipe, a temperature of 0.4 ok was obtained. This indicates that He2 cryostats can be considerably simplified by using absorption pumps only. Acknowledgments are There are 1 figure and 14 references: 6 Soviet-bloc and 8 nonexpressed to B.G. Lazarev for his advice. Soviet-bloc. The four most recent English language references read as follows: Ref. 8: G. Seidel, P.H. Keesom, Rev. Scient, Instrum., v.29, 1958, 606, Ref. 10: H.A. Reich, R.L. Garwin, Rev. Scient. Instrum., v.30, 1959, 7. Card 3/8 4

33155

An  $\text{He}^3$  apparatus for the production...  $\frac{\text{S}/120/61/000/006/026/041}{\text{E032/E114}}$ 

Ref. 13: C.J.N. v. d. Meydenberg, K.W. Taconis, 7th Intern. Conf. on Low Temp. Phys., Toronto, Programme, 1960.
Ref. 14: as in text above.

ASSOCIATION: Fiziko-tekhnicheskiy institut AN USSR (Physicotechnical Institute, AS Ukr.SSR)

SUBMITTED: January 25, 1961

Card 4/# 4

YESEL'SON, B.M.; IAZAREV, B.G.; SHVETS, A.D.

Obtaining lower than 1° K. temperatures by pumping-off liquid helium vapors with an adsorption pump. Prib.i tekh.eksp. 6 no.5:160-162 S-0 '61. (MIRA 14:10)

1. Fiziko-tekhnicheskiy institut AN USSR. (Low temperature engineering)

YESEL'SON, B.N.; SHVETS, A.D.; BEREZNYAK, N.O.

Device for obtaining temperatures up to 0,30 K. using He<sup>3</sup>. Prib. i tekh.eksp. 6 no.6:123-124 K-D '61. (MIRA 14:11)

1. Fiziko-tekhnicheskiy inetitut AN USSR. (Cryostat)

5/120/62/000/003/048/048 E032/E114 Yesel'son, B.N., Lazarev, B.G., and Shvets, A.D. 9,5110 AUTHORS: PERICUICAL: Pribory i tekhnika eksperimenta, no.3, 1962, 198-199 A simple He3 cryostat It is pointed out that existing He3 cryostats capable of producing temperatures down to 0.3 K are rather complicated because they incorporate diffusion pumps and/or rotary pumps to pump the vapour above liquid helium and thereby reduce the temperature. The present authors have used a charcoal adsorption pump to remove the vapour and thereby have simplified the construction and succeeded in producing temperatures down to 0.34 °K. The device is shown in the figure, in which: 1 - charcoal pump; 2 - thin-walled stainless steel tube; 1 - cnarcoal pump; 2 - thin-walled stainless steel tube;
3 - reservoir containing He3; 4, 10 - cylinders for storing
helium gas; 5 - vacuum jacket; 6 - valve connecting the charcoal
pump 1 to the reservoir 3; 7 - dewar with liquid helium at
pump 1 to the reservoir 5; 7 - dewar with He3 - He4 solution
1.3 °K; 8 - container filled either with He3 - He4 solution 1.3 °K; 8 - container filled either with He3 - He4 solution (7.4% He3) or pure He4; 9 - not given; 11 - tube for removing helium gas. Card 1/0 7

A simple He<sup>3</sup> cryostat

S/120/62/000/003/048/048 E032/E114

11

The charcoal pump is in the form of a brass cylinder containing about 30 g of activated charcoal which is held in position by a pair of brass grids. The temperature of 0.34 °K is reached after about 30 minutes. The heat leak of the system is about 600 erg/sec.

There is I figure.

ASSOCIATION: Fiziko-tekhnicheskiy institut, AN USSR

(Physicotechnical Institute, AS Ukr.SSR)

SUBMITTED: November 29, 1961

Card 2/8 7

37097 s/056/62/042/004/003/037

AUTHORS:

Yesel'son, B. N., Ivantsov, V. G., Shvets, A. D.

The A-roint of concentrated He3-He4 solutions

PERIODICAL:

Zhurnal eksperimental noy i teoreticheskoy fiziki, v. 42,

no. 4, 1962, 944-948

TEXT: The authors continue earlier investigations (ZhETF,  $\underline{20}$ , 748, 1950; DAN SSSR,  $\underline{111}$ , 568, 1956; ZhETF,  $\underline{31}$ , 902, 1956; ZhETF,  $\underline{34}$ ,  $\underline{233}$ , 1958) of the He I  $\rightarrow$  He II transition point ( $\underline{T}_{\lambda}$ ) as dependent on the He<sup>3</sup> concentration (X). The  $\underline{T}_{\lambda}(X)$  dependences were then determined for higher He<sup>3</sup> concentrations (50.0, 59.6, 62.4%).  $\underline{T}_{\lambda}$  of the He-solution with known He<sup>3</sup> content was determined from the particularities of the heating of content was determined from the particularities of the heating or cooling rate curves which were recorded by an ANN-09 (EPP-09) electronic potentiometer. The measurements were carried out in an apparatus consisting of several Dewar vessels in which temperatures below 10K could be reached by pumping out the vapor above the liquid He<sup>4</sup> by an adsorption pump. For the above He<sup>3</sup> concentrations the T, values were 1.31 ± 0.01°K, pump. 1.05 ± 0.01°K and 1.02 ± 0.02°K pump. For the above her concentrations the  $1_{\lambda}$  varies age 1.01  $\pm$  0.01 K and 1.02  $\pm$  0.03 K. For a solution with X = 66.1%,  $T_{\lambda}$  could Card 1/2

The  $\lambda$ -point of concentrated ...

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not be determined. The values obtained are shown in a  $T_{\lambda}(X)$  graph together with data of many other publications. The data fit a curve which is almost a straight line. Professor B. G. Lazarev is thanked for discussions and V. D. Krasnikov for assistance. There are 4 figures.

ASSOCIATION: Fiziko-tekhnicheskiy institut Akademii nauk Ukrainskoy SSR

(Physicotechnical Institute of the Academy of Sciences

Ukrainskaya SSR)

SUBMITTED:

September 20, 1961

Card 2/2

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#### "APPROVED FOR RELEASE: 03/15/2001 CIA-

CIA-RDP86-00513R001962920004-8

43383 \$/056/62/043/005/056/058 B125/B104

//, 3120 AUTHORS: Bereznyak, N. G., Bogoyavlenskiy, I. V., Yesel'son, B. N.

TITLE:

The curves representing the onset of solidification of helium

isotope solutions

PERIODICAL:

Zhurnal eksperimental'noy i teoreticheskoy fiziki, v. 43,

no. 5(11), 1962, 1981-1982

TEXT: The method of thermal analysis was used to establish a correlation between the solidification pressure and the composition of the liquid phase in order to draw the diagram for the equilibrium between the solid and the liquid phase of solutions of He<sup>3</sup> in He<sup>4</sup>. The temperature and pressure at which the solutions of He<sup>3</sup> in He<sup>4</sup> begin to solidify (10.3; 24.1; 53.0 and which the solutions of He<sup>3</sup> in He<sup>4</sup> begin to solidify (10.3; 24.1; 53.0 and 76.4%)He<sup>3</sup>) can be determined from the salient points of the curve representing the time dependence on temperature and pressure. A resistance thermometer was used to measure the temperature of the calorimeter, whilst thermometer was used to measure the temperature of the calorimeter, whilst the pressure inside the latter was determined from the elastic deformation the calorimeter wall, using a strain gauge. Between 1.5 and 4.20K, the

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The curves representing...

pressure at the beginning of liquefaction increases as the portion of He3 increases in the solution (Fig. 1). The dependence of the solidification pressure on the He3 portion in the solution is constructed from these data at various temperatures (Fig. 2). The shape of the isotherms, and the good agreement with the results obtained by blocking the capillary tubes, are indicative of a narrow "demixing region" in the above-mentioned equilibrium diagram. The present results agree satisfactorily with recent data obtained for the temperature range from 1.0 to 2.10K. The point at which solutions of He in He cease to solidify is now being determined. There are 2 figures.

ASSOCIATION:

Fiziko-tekhnicheskiy institut Akademii nauk Ukrainskoy SSR (Physicotechnical Institute of the Academy of Sciences of the

Ukrainskaya SSR)

SUBMITTED:

September 12, 1962

Card 2/4